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| APPLICATION NO. | FILING DATE | FIRST NAMED INVENTOR | ATTORNEY DOCKET NO. | CONFIRMATION NO |
|---|-------------|----------------------|---------------------|-----------------|
| 10/542,770 | 07/17/2006 | Roger Scattergood | K0181.70020US00 | 1614 |
| 29428 7590 09(15/2008 WOLF GREENFIELD & SACKS, P.C. 600 ATLANTIC AVENUE | | | EXAMINER | |
| | | | HINES, LATOSHA D | |
| BOSTON, MA 02210-2206 | | | ART UNIT | PAPER NUMBER |
| | | | 1797 | |
| | | | | |
| | | | MAIL DATE | DELIVERY MODE |
| | | | 09/15/2008 | PAPER |

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Application No. Applicant(s) 10/542,770 SCATTERGOOD, ROGER Office Action Summary Examiner Art Unit LATOSHA HINES 1797 -- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --Period for Reply A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS. WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION. Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication. If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication - Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b). Status 1) Responsive to communication(s) filed on July 17, 2006. 2a) ☐ This action is FINAL. 2b) This action is non-final. 3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under Ex parte Quayle, 1935 C.D. 11, 453 O.G. 213. Disposition of Claims 4) Claim(s) 1-27 and 29-41 is/are pending in the application. 4a) Of the above claim(s) is/are withdrawn from consideration. 5) Claim(s) _____ is/are allowed. 6) Claim(s) 1-27 and 29-41 is/are rejected. 7) Claim(s) _____ is/are objected to. 8) Claim(s) _____ are subject to restriction and/or election requirement. Application Papers 9) The specification is objected to by the Examiner. 10) ☐ The drawing(s) filed on is/are: a) ☐ accepted or b) ☐ objected to by the Examiner. Applicant may not request that any objection to the drawing(s) be held in abevance. See 37 CFR 1.85(a). Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d). 11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152. Priority under 35 U.S.C. § 119 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f). a) All b) Some * c) None of: Certified copies of the priority documents have been received. 2. Certified copies of the priority documents have been received in Application No. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). * See the attached detailed Office action for a list of the certified copies not received. Attachment(s)

1) Notice of References Cited (PTO-892)

Paper No(s)/Mail Date 07/20/2005.

2) Notice of Draftsperson's Patent Drawing Review (PTO-948)

Interview Summary (PTO-413)
Paper No(s)/Mail Date.

6) Other:

Notice of Informal Patent Application

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DETAILED ACTION

This is the initial Office Action based on the 10/542770 application filed on July
2006.

 Claims 1-42 are pending and have been fully considered. Claims 28 and 42 have been canceled through amendment to claims.

Claim Rejections - 35 USC § 102

3. The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

 Claims 1-13, 26-27, 29, 32, and 41 are rejected under 35 U.S.C. 102(b) as being clearly anticipated by HAZARIKA et al. (WO 02/00812 A2).

With respect to claim 1 HAZARIKA et al. discloses a method for improving the efficiency of combustion processes and/or reducing harmful emissions through composition, tablet, capsule or liquid fuel additive suitable for dispersing a lanthanide (rare earth) oxide in a fuel (page 1 lines 3-6). Preferably, the lanthanide oxide comprises a lanthanide selected from cerium, lanthanum, neodymium and praseodymium. Preferably, the lanthanide oxide is CeO₂ (cerium oxide) (page 4 lines 1-2). The substance used to coat the surface of the lanthanide oxide is preferably a surfactant (detergent fuel additive). The lipophobic part of the surfactant is embedded into the lanthanide oxide particle,

leaving the lipophilic part of the surfactant to interact with the fuel (page 6 lines 2-4).

With respect to claim 2-5 HAZARIKA et al. discloses other materials may be added to the fuel in addition to the lanthanide oxide such as compounds of manganese, iron, cobalt, nickel, barium, strontium, calcium and lithium (page 6 lines 21-27).

With respect to claims 6-7 HAZARIKA et al. discloses particles of lanthanide oxide (cerium oxide) added to the fuel are discrete particles. Preferably, the mean particle size of the lanthanide oxide is in the range of 1 nm to 100 microns. More preferably, the mean particle size is in the range of 1 nm to 5 microns, more preferably 1 nm to 0.5 microns, more preferably 1 nm to 50 nm, and more preferably 1 nm to 10 nm (page 4 lines 9-14).

With respect to claim 8-11 HAZARIKA et al. discloses surfactants having a low HLB generally more oil soluble than those surfactants having a high HLB. Examples of low HLB surfactants are alkyl carboxylic acids, **anhydrides and esters** having at least one C₁₀-C₃₀ alkyl group, such as **dodecenyl succinic anhydride (DDSA)** (dicarboxylic acid anhydride/alkenyl succinic anhydride), stearic acid, oleic acid, sorbitan tristearate and glycerol monostearate (page 6 lines 5-14).

With respect to claim 12 HAZARIKA et al. discloses an internal combustion engine which may be any type including spark ignition engines and compression ignition engines. Similarly, the fuel may be of any type, including petrol/gasoline (both leaded and unleaded), **diesel** and LPG fuel (page 3 lines 2-3).

With respect to claim 13 HAZARIKA et al. discloses an **aromatic hydrocarbon** compound, such as benzene or naphthalene, optionally substituted with one or more C₁-C₆ alkyl group(s) (page 8 lines 2-4).

With respect to claim 17-18 HAZARIKA et al. discloses the substance used to coat the surface of the lanthanide oxide is preferably a surfactant (detergent). The lipophobic part of the surfactant is embedded into the lanthanide oxide particle, leaving the lipophilic part of the surfactant to interact with the fuel (page 6 lines 2-4).

With respect to claim 26 and 27 HAZARIKA et al. discloses the amount of

lanthanide oxide required will depend on the total surface area of the lanthanide oxide particles and also fuel tank capacity. Preferably, the amount of lanthanide oxide added to the fuel is 1 to 10 ppm (page 5 lines 2-3 and line 10). With respect to claim 29 HAZARIKA et al. discloses the substance used to coat the surface of the lanthanide oxide is preferably a surfactant (detergent). The

lipophobic part of the surfactant is embedded into the lanthanide oxide particle, leaving the lipophilic part of the surfactant to interact with the fuel (page 6 lines 2-4).

With respect to claim 32 HAZARIKA et al. discloses a lanthanide oxide comprising a lanthanide selected from cerium, lanthanum, neodymium and praseodymium. Preferably, the lanthanide oxide is CeO₂ (cerium oxide) (page 4

lines 1-2). The substance used to coat the surface of the lanthanide oxide is preferably a surfactant (detergent fuel additive). The lipophobic part of the surfactant is embedded into the lanthanide oxide particle, leaving the lipophilic part of the surfactant to interact with the fuel (page 6 lines 2-4). Other materials may be added to the fuel in addition to the lanthanide oxide such as compounds of manganese, iron, cobalt, nickel, barium, strontium, calcium and lithium (page 6 lines 21-27).

With respect to claim 41 HAZARIKA et al. discloses an aromatic hydrocarbon compound, such as benzene or naphthalene, optionally substituted with one or more C₁-C₆ alkyl group(s) (page 8 lines 2-4).

Claim Rejections - 35 USC § 103

- The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
 - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- Claims 14-16 and 30-31 are rejected under 35 U.S.C. 103(a) as being unpatentable over <u>HAZARIKA et al. (WO 02/00812 A2)</u>.

With respect to claim 14-16 HAZARIKA et al. discloses the lanthanide oxide may be in the form of a loose powder, tablet, and capsule or liquid fuel additive.

These may be dispensed into fuels manually (e.g. by addition to the fuel tank at the time of refueling) or with the aid of a suitable mechanical or electrical dosing device that may be utilized to automatically dose an appropriate amount of

lanthanide oxide into the fuel (page 9 lines 13-17). It would have been obvious to add the compound at either of these locations for ease of handling the cerium additive (adding a concentrate) and to ensure thorough mixing of the fuel and additive before introduction of the fuel to the vehicle.

With respect to claims 30 and 31 HAZARIKA et al. discloses the amount of lanthanide oxide is in the range of 1 to 99.99 wt% based on the total weight of the tablet. More preferably, the amount of lanthanide oxide is in the range of 30 to 80 wt% and more preferably 40 to 60 wt. %. More preferably, the amount of lanthanide oxide in the tablet is about 50 wt. % (page 8 lines 6-9). The discovery of an optimum value of a known result effective variable, without producing any new or unexpected results, is within the ambit of a person of ordinary skill in the art. See *In re Boesch*, 205 USPQ 215 (CCPA 1980) (See MPEP 2144).

 Claims 19-25 and 33-40 are rejected under 35 U.S.C. 103(a) as being unpatentable over <u>HAZARIKA et al. (WO 02/00812 A2)</u> in view of <u>COLLIER et al. (US 2003/0182848)</u>.

With respect 19-25 and 33-40 HAZARIKA et al. does not teach a cerium oxide wherein a detergent is a succinimide derived from a polyisobutenyl succinic acylating agent and an ethylene polyamine having an average composition from triethylene tetramine to pentaethylene hexamine.

However, with respect to claim 19-21 and 33-35 COLLIER et al. discloses a metal salt-ashless detergent additive combination prepared by combining the aforesaid esters of C_2 - C_{50} carboxylic acids with an ashless dispersant comprising

an acylated nitrogen compound having a hydrocarbyl substituent of at least 10 carbon atoms made by reacting an acylating agent with an amino compound. such as the reaction products of polyisobutenyl succinic anhydride wherein the **poly(isobutene)** substituent has between about 50 to about 400 carbon atoms with a mixture of ethylene polyamines having 3 to 7 amino nitrogen atoms (paragraph 0080 and 0100). The discovery of an optimum value of a known result effective variable, without producing any new or unexpected results, is within the ambit of a person of ordinary skill in the art. See In re Boesch, 205 USPQ 215 (CCPA 1980) (See MPEP 2144). With respect to claim 22-24 and 36-38 COLLIER et al. discloses a preferred acylated nitrogen ashless detergent compounds are those made by reacting a poly(isobutene) substituted succinic anhydride acylating agent with mixtures of ethylene polyamines such as lower alkylene diamines, ethylene diamine (paragraphs 0078 and 0086). With respect to claim 25 and 39-40 COLLIER et al. discloses diesel fuel compositions can contain other additives which are well known to those of skill in the art. These include dyes, cetane improvers, rust inhibitors, bacteriostatic agents, gum inhibitors, metal deactivators, demulsifiers, upper cylinder lubricants and antiicing agents and antioxidants (paragraph 0098).

At the time of the invention it would have been obvious to one of ordinary skill in the art to use the succinimide and various fuel additives of COLLIER et al. in the fuel composition of HAZARIKA et al. because the succinimide and various fuel additives are stable systems that are cooperatively effective in fuel in improving

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the operation of diesel engine particulate traps as taught by COLLIER et al (paragraph 0014).

Therefore, the invention as a whole would have been prima facie obvious to one of ordinary skill in the art at the time the invention was made.

Conclusion

 Any inquiry concerning this communication or earlier communications from the examiner should be directed to LATOSHA HINES whose telephone number is 571-270-5551. The examiner can normally be reached on Monday thru Thursday from 8 a.m. to 5 p.m.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Glenn Caldarola can be reached on 571-272-1444. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only.

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/LATOSHA HINES/ Examiner, Art Unit 1797

//Cephia D. Toomer// Primary Examiner, Art Unit 1797